## ROBINSON SECONDARY SUMMER 2024 Activity IB Chemistry Standard Level

**Purpose:** Welcome to Standard Level IB Chemistry! The purpose of this activity is to review material from IB Chemistry 1 until it is mastered. Mastery of the material covered in IB Chemistry I is absolutely essential for success in IB SL Chemistry. I will be available by email to assist you while you review the material from the first year to prepare for an exciting year in SL Chemistry.

## **Directions:**

- 1. Find your Chemistry notebook from IB Chemistry 1. You will want to use your notes, objectives and resources to help you.
- 2. Log on to <a href="https://sites.google.com/fcpsschools.net/sl-chemistry-at-robinson/home">https://sites.google.com/fcpsschools.net/sl-chemistry-at-robinson/home</a>, you will find your Summer Assignment.
- 3. The IB Chemistry Textbook used for this assignment is located as its own tab. You will need to complete the following assignments from this IB Chemistry Textbook. Multiple choice questions should be solved with the work shown but may be printed out. You do <u>NOT</u> have to copy all of the questions word for word. You <u>MAY</u> also print charts and fill in the blanks to help make this assignment easier, but printing is <u>NOT</u> required.
  - a. Chapter 1 Quantitative Chemistry
    1.2.3 pg 7-8 #1-9 (odd only)
    1.2.4 pg12-13 #11,13,17
    1.3 pg 14 #5
    1.3.1 pg 17 #4 (a-c), 5(a-c)
    1.4.1 pg 21 #2
    1.4.2 pg 24 #6-8
    1.4.3 pg 26 #6-8
    - 1.5 pg 29 #7-9 1.5.1 pg31 #4
  - b. Chapter 2 Atomic Structure 2.1 pg 50-52 #1-12
  - c. Chapter 3 Periodicity
    3.1 pg 72 #1-3
    3.2 pg 75 #1-6
  - d. Chapter 4 Chemical Bonding4.1 pg 101 #1-54.2 pg 106 #1-6, 7 a-c, 8
  - e. Chapter 5 Energetics 5.1 pg 137 #1-4
  - f. Chapter 11 Measurement and Data 11.1 pg 298 #1-6

Polyatomic Ions					
Formula	Name				
C <sub>2</sub> H <sub>3</sub> O <sub>2</sub>	ethanoate (acetate)				
CN-	cyanide				
SCN <sup>-</sup>	thiocyanate				
H <sub>2</sub> PO <sub>4</sub> -	dihydrogen phosphate				
HCO <sub>3</sub>	hydrogen carbonate (bicarbonate)				
HSO <sub>4</sub>	hydrogen sulfate (bisulfate)				
OH.	hydroxide				
NO <sub>3</sub> -	nitrate				
NO <sub>2</sub>	nitrite				
MnO <sub>4</sub> -	permanganate				
CO3 <sup>2-</sup>	carbonate				
CrO <sub>4</sub> <sup>2</sup> ·	chromate				
Cr <sub>2</sub> O <sub>7</sub> <sup>2</sup> ·	dichromate				
HPO <sub>4</sub> <sup>2</sup> ·	hydrogen phosphate				
SO4 <sup>2-</sup>	sulfate				
SO <sub>3</sub> <sup>2-</sup>	sulfite				
PO <sub>4</sub> 3·	phosphate				
NH <sub>4</sub> <sup>+</sup>	ammonium				

- 4. **REVIEW AND MEMORIZE the polyatomic ions listed on this assignment**. They are also posted under course materials on our summer assignment Blackboard site. You should know the name, charge, and formula.
- 5. **REFLECT on an Individual Assessment Investigation Project** (think individual science fair project on a Chemistry topic you could do here at school). Your proposal should include the IV (levels, units and how it will be measured), DV (how it will be measured), hypothesis with explanation and 1 paragraph of background research (with references). Format and form can be found on Summer Assignment Google Site. Please consider scope of project and safety as you are looking at possible ideas. Contact me if you have and questions or concerns.

Working on this assignment and reviewing your Chemistry 1 material will help you start the year strong. This assignment is not graded but an answer key will be available at the start of the year. I can be contacted over the summer by email - dagross@fcps.edu. Please send me your full name when emailing me with questions.